

## Metabolism of Labeled Organic Nitrogen in Soil: Regulation by Inorganic Nitrogen

M. Scott Smith,\* Charles W. Rice, and Eldor A. Paul

### ABSTRACT

Regulation of organic N metabolism by inorganic N availability was investigated in short-term laboratory incubations of soil. A  $^{14}\text{C}$ -,  $^{15}\text{N}$ -labeled organic N substrate was produced by growing *Pseudomonas stutzeri* in labeled media and isolating a cytoplasmic fraction. This was added to soils that had been preincubated with glucose or glucose plus  $\text{NH}_4^+$  to induce conditions of N deficiency or sufficiency. Regulation by inorganic N was indicated by stimulated proteolytic enzyme activity and greater initial rates of cytoplasmic  $^{14}\text{C}$  mineralization in N deficient soils. However, effects of N deficiency on  $^{14}\text{C}$  mineralization persisted for no more than 24 h. Preinduced N deficiency significantly decreased the extent of  $^{15}\text{N}$  mineralized from cytoplasmic N. Mineralization of  $^{14}\text{C}$  from leucine added to soil was similarly affected by N availability, yet  $^{14}\text{C}$ -glutamate mineralization was apparently unaffected. In another experiment labeled cytoplasm was added simultaneously with or without a larger quantity of glucose. The glucose caused virtually complete assimilation of  $^{15}\text{N}$  but had no effect on apparent assimilation of  $^{14}\text{C}$ . Thus, there was no relationship between  $^{15}\text{N}$  assimilation and  $^{14}\text{C}$  assimilation, suggesting that the C and N contained in organic N are processed separately by soil microbes. Inorganic N availability may have short-term effects on metabolism of C in organic N but long-lasting effects appear to be minimal.

THE REGULATORY function of inorganic N ( $N_i$ ) in the metabolism of organic C and organic N ( $N_o$ ) in soil systems is not clear. Where  $N_i$  additions have stimulated C mineralization ( $\text{CO}_2$  evolution) from organic materials this probably can be attributed to continuous, extreme N deficiency limiting microbial growth (Jenkinson et al., 1985). More frequently  $N_i$  amendment has inconsistent effects (Johnson et al.,

1980; Shields et al., 1974) or inhibits C mineralization (Kowalenko et al., 1978; De Jong et al., 1974). In many of these studies it is not possible to assess the overall N and C (energy) availability in the unamended soils; this is likely to influence the relative response to  $N_i$  amendment. In addition to uncertainty about  $N_i$  effects on the rate of C mineralization, there is the question of  $N_i$  effects on the extent of C assimilation and mineralization, and therefore the significance of N availability in determining soil organic matter retention (Leuken et al., 1962; Pinck et al., 1950; Turk and Millar, 1936).

With regard to metabolism of  $N_o$ , pure culture results suggest that regulation by  $N_i$ , specifically  $\text{NH}_4^+$ , will be significant. In the fungi and bacteria that have been characterized, enzymes for utilization of C-N compounds tend to be regulated by both C and N availability (Marzluf, 1981; Payne, 1980; North, 1982). Yet, the growing fundamental understanding of regulation of amino acid and protein utilization obtained in pure cultures is difficult to extrapolate to soil systems. In part, this is because regulatory mechanisms and the specific effects of  $\text{NH}_4^+$  vary among organisms. Also,  $N_o$  utilization requires distinct enzymatic steps for extracellular hydrolysis, uptake, deamination, and intracellular catabolism, each of which could be regulated differently. Although proteolytic enzyme activity and degradation of  $N_o$  have been studied in soil (Ladd, 1978), the possible significance of  $N_i$  regulation apparently has not been considered.

Some simulation models of C-N cycling in soils have accounted for the possibility of regulation by  $N_i$  and other nutrients by including demand or deficiency factors, related to nutrient availability, in the substrate decomposition functions (Bosatta and Berendse, 1984; Van Veen et al., 1981). The latter authors specifically noted that it is unclear how regulation of extracellular polymer metabolism should be described mechanistically. In conceptual models (McGill and Cole, 1981; Hunt et al., 1983) it has been proposed that soil nutrients covalently bonded with C (C-N) are metabolized to provide energy whereas ester bonded nutrients (C-O-P or C-O-S) are metabolized to provide the

M.S. Smith, Dep. of Agronomy, Univ. of Kentucky, Lexington, KY 40546; C.W. Rice, Dep. of Agronomy, Kansas State Univ., Manhattan, KS 66506, formerly Dep. of Crop and Soil Sciences, Michigan State Univ., and E.A. Paul, Dep. of Crop and Soil Sciences, Michigan State Univ., E. Lansing, MI 48824. The investigation reported in this paper (no. 88-3-117) is in connection with a project of the Univ. of Kentucky Agric. Exp. Stn. and is published with the approval of the director. The project was initiated while M.S. Smith and E.A. Paul were at the Dep. of Plant and Soil Biology, Univ. of California, Berkeley, and was supported in part by NSF grant B.S.R. 8306181. Received 8 Aug. 1988. \*Corresponding author.

nutrient. If this is true, metabolism of C-N compounds should be regulated primarily by C availability and not  $N_i$ . Yet, this is inconsistent with pure culture studies as discussed above.

We have used  $^{14}\text{C}$ -,  $^{15}\text{N}$ -labeled bacterial cytoplasm and  $^{14}\text{C}$ -labeled amino acids to determine how the availability of  $N_i$  affects the metabolism of easily degraded sources of  $N_o$ .

## MATERIALS AND METHODS

### Soils

Two soils were used in this study. An Argonaut silt loam (Mollic Haploxeralfs) was collected from the oak-grass savannah on the Sierra Nevada Foothill Range Stn., Univ. of California. This soil has a pH of 5.9, and organic C and N contents of 31 g C  $\text{kg}^{-1}$  and 2.2 g N  $\text{kg}^{-1}$ . The second soil was a Holland silt loam (Ultic Haploxeralfs) collected from a mixed conifer forest on the Blodgett Exp. For., Univ. of California. This soil has a pH of 5.6 and organic C and N contents of 40 g C  $\text{kg}^{-1}$  and 2 g N  $\text{kg}^{-1}$ .

### Production of Labeled Organic N

*Pseudomonas stutzeri* was grown aerobically in media which included (per liter): 0.2 g yeast extract, 0.2 g  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , 0.68 g  $\text{KH}_2\text{PO}_4$ , 1.14 g  $\text{K}_2\text{HPO}_4 \cdot 3\text{H}_2\text{O}$ , 0.10 g  $\text{NH}_4\text{Cl}$ , 2 g glucose, 5 mg  $\text{Na}_2\text{EDTA}$ , 5 mg  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ , 1.5 mg  $\text{MnSO}_4 \cdot \text{H}_2\text{O}$ , 0.1 mg  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ , 0.2 mg  $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ , 0.18 mg  $\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$ , and 0.58 mg  $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$ . For labeled batches, uniformly labeled  $^{14}\text{C}$  D-glucose, 75  $\mu\text{Ci l}^{-1}$ , was used and  $\text{NH}_4\text{Cl}$  was replaced with 237.7 mg  $(\text{NH}_4)_2\text{SO}_4$  enriched with  $^{15}\text{N}$ .

Cells were harvested in early stationary phase by centrifugation (4 °C, 10 000 g). These were washed repeatedly in pH 7 phosphate buffer, resuspended in water, then disrupted by sonication. A fraction consisting primarily of cell walls was removed by centrifugation. The supernatant was decanted, and acidified to pH 3 with HCl. This precipitated cytoplasmic protein and presumably other cytoplasmic polymers. Soluble components were removed from this precipitate by repeated centrifugation and washing in dilute HCl. The precipitated  $N_o$  was resolubilized by neutralizing with KOH, then frozen for storage. This material had a C/N ratio of approximately 4/1. The batch used for experiment I had an atomic %  $^{15}\text{N}$  of 6.305; for Experiment II and IV this was 17.647.

### Experiment I

Argonaut soil (35.1 g dry wt) was added to 945 mL Mason jars. This was amended with glucose, glucose plus  $\text{NH}_4\text{Cl}$  or with nothing. There were four replicates of each of these three treatments. The concentration of glucose was 1.92 mg C g soil $^{-1}$ ;  $\text{NH}_4\text{Cl}$  was 0.19 mg N g $^{-1}$  soil. The final moisture content of all treatments was 0.28 g  $\text{H}_2\text{O g}^{-1}$  dry soil (0.08 MPa). The jars were sealed with covers fitted with a serum stopper for gas sampling. Unamended initial soil samples were analyzed for  $N_i$  ( $\text{NH}_4^+$  and  $\text{NO}_3^-$ ). Periodically gas samples from the jars were analyzed for  $\text{CO}_2$ . When  $\text{CO}_2$  concentration exceeded 2% by volume the covers were removed briefly to replenish the  $\text{O}_2$  supply. Samples were incubated at 24 °C in the dark.

After 7 d, a subsample (5 g) was removed for  $N_i$  analysis. The remaining soil was then amended with 2 mL of labeled  $N_o$  solution containing 26.9  $\mu\text{g N}$  and 98.4  $\mu\text{g C}$  per g dry soil. Periodically during the next 4 d headspace gas samples were collected by syringe for measurement of total and  $^{14}\text{C}$ - $\text{CO}_2$ . On day 11, subsamples (5 g) were removed for measurement of total  $N_i$ .

### Experiment II

This was similar in design to experiment I with exceptions described below. Holland soil (15.3 g dry wt) was added to 473-mL Mason jars. All flasks received 1.96 mg glucose-C g $^{-1}$  soil. Half of the flasks received 0.196 mg  $\text{NH}_4\text{Cl-N g}^{-1}$  soil. There were eight replications of each treatment. Moisture content after amendment was 0.31 g  $\text{H}_2\text{O g}^{-1}$  dry soil (0.03 MPa). The preincubation period lasted for 6 d. At this time four replicates were used for  $N_i$  measurement; the remaining jars were amended with labeled  $N_o$  containing 58.2  $\mu\text{g N}$  and 232.7  $\mu\text{g C g}^{-1}$  soil and were incubated for 4 additional days.

### Experiment III

Argonaut soil (31 g dry wt.) was placed in 945-mL Mason jars and amended with 1.9 mg glucose-C g $^{-1}$ , with 0.19 mg  $\text{NH}_4\text{Cl-N g}^{-1}$  plus glucose, or left unamended. The final gravimetric moisture content was 0.28 g  $\text{H}_2\text{O g}^{-1}$  dry soil (0.08 MPa). There were three replications per treatment. Jars were incubated as above. Periodically subsamples were removed for measurement of proteolytic activity as described below.

### Experiment IV

Argonaut soil (14.8 g dry wt) was incubated in 473-mL Mason jars. Half the samples were amended with 2.03 mg glucose C g $^{-1}$  soil and all received, at the same time, labeled cytoplasm at a rate of 0.11 mg C g $^{-1}$  and 0.027 mg N g $^{-1}$ . Soils were incubated as above. There were four replicates per treatment. Total  $\text{CO}_2$ ,  $^{14}\text{CO}_2$ ,  $N_i$ , and  $^{15}\text{N}_i$  were measured at 228 h. Initial  $N_i$  was also determined.

### Analytical Procedures

Proteolytic activity was measured by determining the rate of production of Folin reagent-reactive compounds from casein, following the procedure of Ladd and Butler (1972). Moist soil (1 g) was mixed with 5 mL of TRIS buffer (pH 8.1, 0.1 M) containing 50 mg isoelectric casein. This was incubated in a shaking water bath at 45 °C for 2.5 h. The reaction was terminated with 2 mL of 17.5% (wt/wt) trichloroacetic acid (TCA). After centrifugation, 2 mL of supernatant was mixed with 3 mL 2.8 M  $\text{Na}_2\text{CO}_3$  and 0.33 mL Folin reagent. After 30 min absorbance was read at 700 nm. Tyrosine in an equivalent solution of TRIS buffer and TCA was used as the standard. Controls for each treatment were incubated with casein omitted. Both this value and the value for a control with soil omitted were subtracted from the appropriate sample values.

Inorganic N was extracted by shaking soil for 1 h with 2 M KCl, 10 mL g $^{-1}$  soil. Ammonium and nitrate in filtered extracts were measured by flow injection analysis (Lachat Quikchem, Mequon, WI). When  $^{15}\text{N}$  was to be measured an aliquot of the extract was steam distilled (Keeney and Nelson, 1982). The still was washed with 95% ethanol and 0.04 M acetic acid between samples. Distillate was collected in excess  $\text{H}_2\text{SO}_4$  and evaporated at 90 °C in an  $\text{NH}_4^+$ -free oven. Mass spectrometer analyses were performed by Isotope Services Inc. (Los Alamos, NM).

Total N and  $^{15}\text{N}$  in the labeled substrate were measured by conventional Kjeldahl methods (Bremner and Mulvaney, 1982), followed by distillation and analysis as above. The substrate contained no detectable free  $\text{NH}_4^+$  or  $\text{NO}_3^-$  prior to digestion.

Total C and  $^{14}\text{C}$  in the substrate was measured by wet oxidation and diffusion (Snyder and Trofymow, 1984). Total C was measured by titration and  $^{14}\text{C}$  was measured by mixing an aliquot of the C trapped in base with 10 mL Scintiverse 2 (Fisher Scientific, Pittsburgh, PA).

Table 1. Total C mineralized ( $\text{CO}_2$  evolved) during preincubation and after organic N addition to Argonaut and Holland soils with different amendments.†

Experiment	Initial amendment g $\text{kg}^{-1}$ soil	Total C mineralized	
		Preincubation	After $N_0$
		g $\text{C} \cdot \text{kg}^{-1}$ soil	
I Argonaut	none	0.17a*	0.16a
	1.92 glucose C	1.12b	0.32b
	1.92 glucose C, +0.19 $\text{NH}_4^+ - \text{N}$	1.04c	0.23c
II Holland	1.96 glucose C	1.10a	0.24a
	1.96 glucose C, +0.20 $\text{NH}_4^+ - \text{N}$	1.03b	0.19b

\* Values within the same column and experiment followed by the same letter are not significantly different at the 0.05 level by pair-wise *t*-tests.

† Preincubation was day 0 to 7 for experiment I; day 0 to 6 for experiment II. Second interval was day 7 to 11 for experiment I; day 6 to 10 for experiment II.

Total  $\text{CO}_2$  evolved during incubation was measured by gas chromatography on a 0.5 mL subsample of the headspace of incubation jars. The gas chromatograph was a Varian 91-P (Varian Associates, Sunnyvale, CA) with a Porapak QS column (Supelco, Bellefonte, PA) at 70 °C and a thermal conductivity detector at 105 °C. The carrier gas was He at a flow rate of 30 mL  $\text{min}^{-1}$ . For determination of  $^{14}\text{CO}_2$ , a 20 mL headspace sample was injected into an evacuated scintillation vial containing 1 mL of 0.5 M NaOH and sealed with a serum stopper. After at least 8 h the vial was opened, 10 mL of Scintiverse 2 was added, and  $^{14}\text{C}$  was measured as above.

#### Experiment V

To further define the controls on  $N_0$  mineralization, amino acids were added to the Argonaut soil as a single known  $N_0$  source. Glucose (2  $\mu\text{g C g}^{-1}$  soil) or  $(\text{NH}_4)_2\text{SO}_4$  (100  $\mu\text{g N g}^{-1}$  soil) were added to the soil, and incubated for 7 d at approximately 0.20 g  $\text{H}_2\text{O g}^{-1}$  soil (0.25 MPa) to create N deficiency or sufficiency, respectively. The  $^{14}\text{C}$ -glutamate or leucine was added to 7.5 g dry weight soil at a rate of 500  $\mu\text{g C g}^{-1}$  soil. This amendment brought the soil to a moisture content of 0.30 g  $\text{H}_2\text{O g}^{-1}$  soil (0.06 MPa). Soil was incubated in 473 mL Mason jars with gas-tight lids fitted with septa for gas sampling. To minimize soil drying during the assay, a beaker of water was placed in each Mason jar. Soil was incubated and preincubated at 25 °C. The jars were sampled (1 mL) for  $\text{CO}_2$  content with a Varian 90-P gas chromatograph with detector and column conditions as described above, except that the column was operated at 35 °C. Determination of  $^{14}\text{CO}_2$  was as described previously except that the scintillant was Safety-Solve (Research Products Int., Mt. Prospect, IL.).

For determination of  $N_i$  and extractable  $^{14}\text{C}$ , 50 mL 2 M KCl was added to the soil, shaken for 1 h, and filtered (Whatman no. 1). The extract was analyzed for  $\text{NH}_4^+$  and  $\text{NO}_2^- + \text{NO}_3^-$  colorimetrically. Extracted  $^{14}\text{C}$  was measured by placing a 1 mL aliquot into a 20 mL scintillation vial containing 10 mL of scintillation cocktail.

## RESULTS AND DISCUSSION

### Preincubation

The intent of treating soil with glucose or glucose plus  $\text{NH}_4^+$  was to establish N deficiency or sufficiency prior to incubating with labeled  $N_0$ . Observations during the preincubation in experiments I and II are presented primarily to define soil conditions when  $N_0$  was

Table 2. Inorganic N during incubation of Argonaut soil with or without glucose, ammonium and organic N (experiment I).†

Amendment	N source	Inorganic N		
		Day 0	Day 7	Day 11
		mg N $\text{kg}^{-1}$ soil		
None	Soil N	57.4a*	63.6a	65.4a
	Labeled N	—	—	9.1b
Glucose	Soil N	57.4a	5.3b	9.8b
	Labeled N	—	—	4.0c
Glucose + $\text{NH}_4^+$	Soil N	248.4b	94.3c	92.1d
	Labeled N	—	—	11.6e

\* Values within a column followed by the same letter are not significantly different at the 0.05 level by pair-wise *t*-tests.

† 26.9  $\mu\text{g N}_0 - \text{N}$  added.

added (Tables 1, 2 and 3). During preincubation  $N_i$  was reduced to low concentrations in soils amended with glucose only. These results also are of interest with regard to the effect of  $N_i$  on metabolism of soil and glucose C. In experiments I and II, total C mineralized was slightly, but consistently and significantly, increased in N deficient (glucose alone) compared to sufficient treatments ( $\text{NH}_4^+$  plus glucose) (Table 1). This  $\text{NH}_4^+$  effect on unlabeled  $\text{CO}_2$  evolution was also observed after  $N_0$  was added (Table 1). In these soils, under these conditions, the glucose added completely disappeared in 3 d or less and consumption was not slower with added  $\text{NH}_4^+$  (data not shown). Therefore, the differences in  $\text{CO}_2$  evolution shown here are not the result of  $\text{NH}_4^+$  inhibition of glucose degradation but are due to increased assimilation of C into biomass or microbial products.

### Proteolytic Enzyme Activity

Proteinase activity, as measured by short-term unlabeled casein hydrolysis in experiment III, was affected by glucose and  $\text{NH}_4^+$  additions (Fig. 1). Proteinase activity was approximately constant during incubation of unamended soil. When glucose plus  $\text{NH}_4^+$  was added, significantly increased activity was observed on day 7; before and after this time there was no significant difference from the control. This is consistent with an earlier report (Ladd and Paul, 1973). When N deficiency was induced by adding only glucose, proteinase activity was greater than both the control and the glucose plus  $\text{NH}_4^+$  treatment on days 4 through 11.

These results indicate that proteolytic activity in soil is regulated by  $N_i$  availability. It is not possible to precisely define the mechanism of regulation. This could be at the cellular level: regulation of extracellular hydrolytic enzyme synthesis and soluble N uptake systems. Also, the N deficient vs. N sufficient incubation conditions probably would induce qualitative and quantitative differences in the soil microbial populations, thereby influencing total soil proteolytic activity. Indirect effects of  $N_i$  on enzyme stability and turnover are also possible.

Enzyme assays of this nature cannot necessarily be extrapolated to the actual mineralization of C and N in  $N_0$  substrates. Therefore, at times corresponding to the activity peaks in Fig. 1, labeled  $N_0$  was added to the soil.

Table 3. Inorganic N during incubation of Holland soil (experiment II).†

Amendment	N source	Inorganic N		
		Day 0	Day 6	Day 10
mg N kg <sup>-1</sup> dry soil				
Glucose	Soil N	44.4a*	3.8a	4.4a
	Labeled N	—	—	13.2b
Glucose + NH <sub>4</sub> <sup>+</sup>	Soil N	240.5b	112.9b	107.5c
	Labeled N	—	—	43.2d

\* Values in the same column followed by the same letter are not significantly different at 0.05 level by pair-wise *t*-tests.

† Glucose plus or minus NH<sub>4</sub><sup>+</sup> was added on day 0, labeled cytoplasm on day 6 (58.2 mg N<sub>o</sub>-N added kg<sup>-1</sup>).

### Mineralization of Labeled N<sub>o</sub>

Preinduced N deficiency increased the initial rate of mineralization of C from the labeled N<sub>o</sub> substrate. This was indicated by greater quantities of <sup>14</sup>CO<sub>2</sub> evolution in Argonaut and Holland soils previously treated to induce N deficiency (Fig. 2 and Table 4, respectively). The effect in the Holland soil was less pronounced but statistically significant. These differences were observed only during a period of 24 h or less after N<sub>o</sub> was added. Effects of N<sub>i</sub> addition cannot be attributed to soil acidity induced by nitrification of added NH<sub>4</sub><sup>+</sup> since variation in pH among all treatments was less than 0.1 unit after the preincubation period (data not shown). These observations support the hypothesis that availability of N<sub>i</sub> regulates soil metabolism of N<sub>o</sub>.

On the other hand, induced N deficiency did not have a long-lasting effect on mineralization of the C contained in N<sub>o</sub>. In experiment I (Fig. 2), 98 h after N<sub>o</sub> was added, 36% of the <sup>14</sup>C added was evolved as CO<sub>2</sub> in the unamended and in the glucose plus NH<sub>4</sub><sup>+</sup> treatments, and 37% in the glucose only treatment. In experiment II (Table 4), after 99 h, 24% was mineralized in the glucose plus NH<sub>4</sub><sup>+</sup> treatment, 25% with glucose only.

The difference in mineralization between the two soils might be explained by a difference in their capacity to adsorb N<sub>o</sub>. This unmeasured potential for

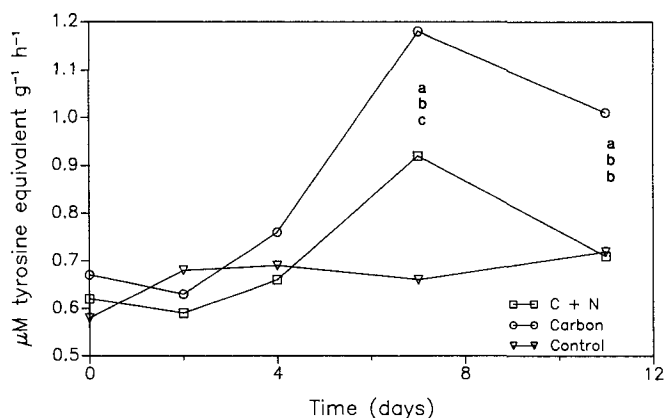


Fig. 1. Proteinase activity, as measured by casein hydrolysis, in Argonaut soil incubated with no amendment, with glucose only, or with glucose plus NH<sub>4</sub><sup>+</sup> (experiment III). Results of statistical analysis are presented in the text. Letters indicate statistically significant differences, at sampling times when they exist, by Duncan's multiple range test (*P* < 0.05).

Table 4. Carbon mineralization following addition of labeled organic N to Holland soil preincubated for 6 d with glucose and with or without NH<sub>4</sub><sup>+</sup> (experiment II).†

Treatment	<sup>14</sup> C recovered as CO <sub>2</sub> after:					
	4 h	9 h	24 h	48 h	75 h	99 h
%						
+NH <sub>4</sub> <sup>+</sup>	1.2a*	3.1a	12.6a	20.0a	23.4a	29.2a
-NH <sub>4</sub> <sup>+</sup>	1.4b	3.6b	10.1b	17.9a	22.2a	25.1a

\* Values in same column followed by same letter are not significantly different at 0.05 level by *t*-tests.

† N<sub>o</sub> added at the rate of 58.2 mg N kg<sup>-1</sup> dry soil.

adsorption also makes it difficult to estimate the fraction of N<sub>o</sub> metabolized which was assimilated into biomass.

Nitrogen deficiency affected total C mineralization differently than it affected labeled C mineralization. While N deficiency caused an initial increase, but no long-lasting effect, on labeled C evolution, it resulted in increased total C evolution throughout the experimental period (Table 1). Unlabeled CO<sub>2</sub> is presumably derived from soil organic matter and from the metabolic turnover of recently added glucose. The results suggest that N regulation of these C sources may be longer-lived, and therefore of greater significance, than N availability effects on N<sub>o</sub> metabolism. Organic N itself had no significant effect on unlabeled C mineralization from N deficient soil, as was determined by incubating the plus glucose minus NH<sub>4</sub><sup>+</sup> treatments with and without N<sub>o</sub> (data not shown).

Nitrogen deficiency had large effects on mineralization of cytoplasmic <sup>15</sup>N, even though the glucose was depleted well before the labeled N<sub>o</sub> was added. In both experiments, approximately three times more of the labeled N was mineralized in the N sufficient treatments compared to the N deficient (glucose only) treatments (Tables 2 and 3). This residual effect of the preincubation on N mineralization might be associated with microbial storage of C reserves or with induced changes in biomass C/N. Significantly less <sup>15</sup>N was mineralized in the previously unamended soil than in the soil with glucose plus NH<sub>4</sub><sup>+</sup> (Table 2). Since the

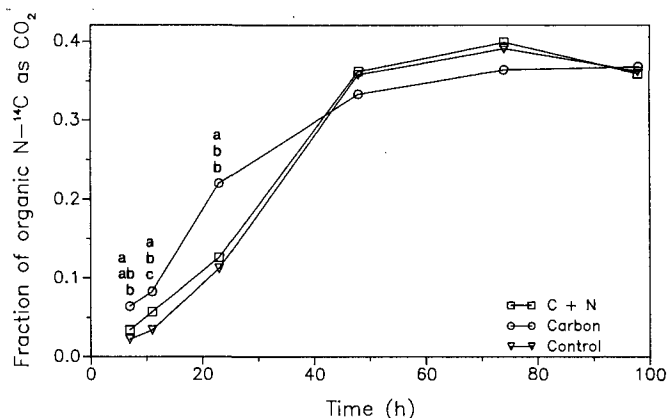


Fig. 2. Mineralization of C from cytoplasmic constituents, as measured by evolution of <sup>14</sup>CO<sub>2</sub> from Argonaut soil preincubated with no amendment, with glucose only, or with glucose plus NH<sub>4</sub><sup>+</sup> (experiment I). Letters indicate statistically significant differences, at sampling times when they exist, by Duncan's multiple range test (*P* < 0.05).

latter had higher concentrations of  $N_i$ , and six times more  $NH_4^+$ , this might be attributed to preferential assimilation of  $NH_4^+$ -N over  $N_o$ .

The fraction of  $N_o$ -N mineralized was higher in experiment II than in experiment I, while the opposite was true for  $N_o$ -C. This argues against adsorption of intact N polymers or amino acids as an explanation for lower C mineralization in experiment II. Approx-

Table 5. Mineralization of C and N from labeled organic N added to Argonaut soil simultaneously with or without glucose and incubated for 9.5 d (experiment IV).†

Treatment	Initial $N_i$	Final $N_i$		C mineralized	
		Total	$^{15}N$	Total	$^{14}C$
mg N kg <sup>-1</sup> soil					
Glucose	31	4	<4‡	1846	45
No glucose	31	61	17	442	46

† Organic N added at the rate of 110 mg C kg<sup>-1</sup> and 27 mg N kg<sup>-1</sup>; glucose added at the rate of 2.03 g kg<sup>-1</sup>.

‡ Because of the small quantity of total  $N_i$  it was not possible to obtain reliable values for  $^{15}N$  mineralized.

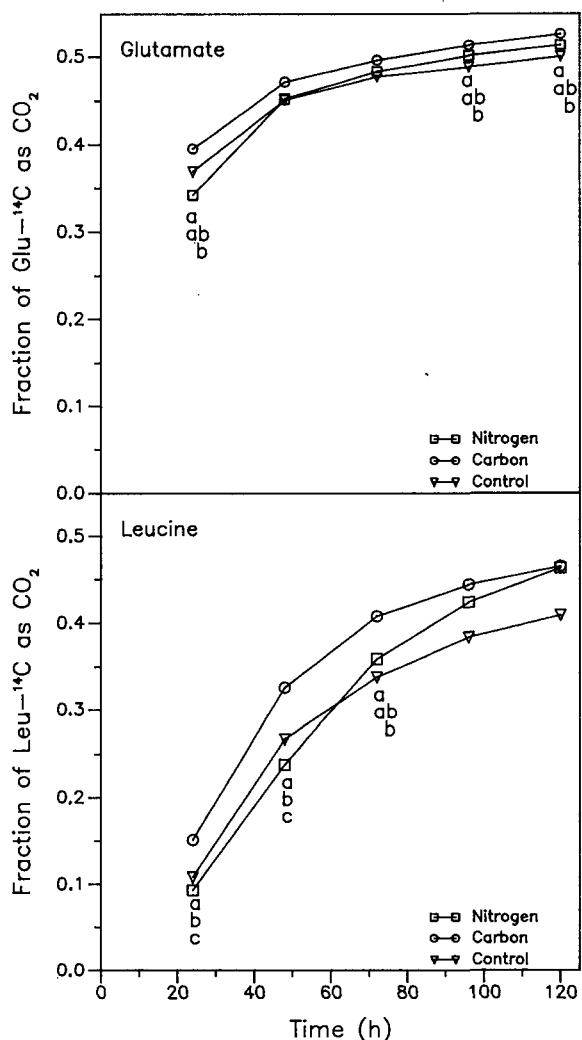


Fig. 3. Rate of C mineralized as  $^{14}CO_2$  from Argonaut soil amended with labeled glutamate or leucine. Soil was preincubated with no amendment, with glucose, or with  $NH_4^+$  (experiment V). Letters indicate statistically significant differences, at sampling times when they exist, by Duncan's multiple range test ( $P < 0.05$ ).

imately twice as much  $N_o$ -N was added in experiment II and there was perhaps a limited potential for N assimilation under these conditions. Yet, the lack of any apparent correlation between C and N assimilation from the same substrate was interesting and this was investigated further in experiment IV.

In experiment IV a relatively small quantity of labeled  $N_o$  was added simultaneously with or without a larger amount of glucose. After 9.5 d, no detectable  $^{15}N$  was mineralized with glucose, but 62% of the  $^{15}N$  was mineralized without glucose (Table 5). Yet there was no significant difference in  $^{14}C$  mineralization from the cytoplasmic material; 41% with glucose, 42% without (Table 5). Therefore in this case, as in the previous experiments, there was no apparent correlation between C assimilation and N assimilation from  $N_o$ . There was a slight delay in  $^{14}C$  mineralization, lasting 2 to 3 d, when glucose was added (data not shown).

### Amino Acids

Glutamate or leucine were added to the Argonaut soil as single defined organic N sources (experiment V). After 3 d incubation, less than 2% of the added glutamate- $^{14}C$ , and 2 to 7% of the leucine- $^{14}C$ , remained in soluble, extractable forms, indicating virtually complete consumption of the initial substrate. Preinduced differences in N availability had no significant effect on the rate of mineralization of glutamate-C (Fig. 3A). In contrast to glutamate, the initial rate of leucine-C mineralization was affected by soil  $N_i$  levels (Fig. 3B). In this case, glucose-induced N deficiency initially increased  $^{14}C$  mineralization relative to the N sufficient treatment. However, this effect did not persist and at the end of the 5-d incubation there was no discernable difference in leucine-C mineralized. Thus, the results with leucine, but not with glutamate, are consistent with those for the cytoplasmic constituents.

### CONCLUSIONS

We had expected that in N deficient soils some of the intact amino acids or nucleic acids from the substrate would be utilized anabolically. It should be energetically advantageous to use preformed amino acids to synthesize cell protein rather than deaminating and catabolizing all substrate amino acids, then synthesizing cell  $N_o$  molecules de novo. Anabolic utilization of intact amino acids would have been reflected in lower  $^{14}C$  mineralization when  $^{15}N$  mineralization was low. Yet this was not the case and there was no relationship between  $^{14}C$  and  $^{15}N$  mineralization and no detectable reduction of C mineralized from  $N_o$  by N deficiency. These results suggest that the C and N in  $N_o$  are processed independently by the soil population. Apparently amino acid and other soluble N substrates are initially deaminated completely then the N and C moieties are apportioned to anabolic and catabolic pathways.

These data provide evidence that  $N_i$  availability can affect  $N_o$  metabolism in soils. This was indicated by increased casein hydrolysis, and by increased initial  $^{14}C$  mineralization from leucine and from cytoplasmic constituents when N deficiency was induced. These

results do not reveal the mechanism for this effect; one possibility is regulation of proteolytic enzyme synthesis by  $N_i$ . However, the effect of  $N_i$  availability was transitory and had little long-term effect on mineralization or assimilation of the C contained in  $N_o$ .

## REFERENCES

- Bosatta, E., and F. Berendse. 1984. Energy or nutrient regulation of decomposition: Implications for the mineralization-immobilization response to perturbations. *Soil Biol. Biochem.* 16:63-67.
- Bremner, J.M., and C.S. Mulvaney. 1982. Nitrogen-total. In A.L. Page et al. (ed.) *Methods of soil analysis*. Part 2. 2nd ed. *Agronomy* 9:595-624.
- DeJong, E., H.J.V. Schappert, and K.B. MacDonald. 1974. Carbon dioxide evolution from virgin and cultivated soil as affected by management practices and climate. *Can. J. Soil Sci.* 54:299-307.
- Hunt, H.W., J.W.B. Stewart, and C.V. Cole. 1983. A conceptual model for interactions among carbon, nitrogen, sulphur and phosphorus in grasslands. p. 303-325. In B. Bolin and R.B. Cook (ed.) *The Major biogeochemical cycles and their interactions*, John Wiley and Sons, Chichester.
- Jenkinson, D.S., R.H. Fox, and J.H. Rayner. 1985. Interactions between fertilizer nitrogen and soil nitrogen—the so called priming effect. *J. Soil Sci.* 36:425-444.
- Johnson, D.W., N.T. Edwards, and D.E. Todd. 1980. Nitrogen mineralization, immobilization and nitrification following urea fertilization of a forest soil under field and laboratory conditions. *Soil Sci. Soc. Am. J.* 44:610-616.
- Keeney, D.R., and D.W. Nelson. 1982. Nitrogen-inorganic forms. In A.L. Page et al. (ed.) *Methods of soil analysis*. Part 2. 2nd ed. *Agronomy* 9:643-698.
- Kowalenko, C.G., K.C. Ivarson, and D.R. Cameron. 1978. Effect of moisture content, temperature and nitrogen fertilization on carbon dioxide evolution from field soils. *Soil Biol. Biochem.* 19:417-423.
- Ladd, J.N. 1978. Origin and range of enzymes in soil. p. 51-96. In R.G. Burns (ed.) *Soil enzymes*. Academic Press, London.
- Ladd, J.N., and E.A. Paul. 1973. Changes in enzymic activity and distribution of acid-soluble, amino acid nitrogen in soil during immobilization and mineralization. *Soil Biol. Biochem.* 5:825-840.
- Ladd, J.N., and J.H.A. Butler. 1972. Short-term assays of soil proteolytic enzyme activities using proteins and dipeptide derivatives as substrates. *Soil Biol. Biochem.* 4:19-30.
- Leuken, H., W.L. Hutcheon, and E.A. Paul. 1962. The influence of nitrogen on the decomposition of crop residues in the soil. *Can. J. Soil Sci.* 42:276-288.
- Marzluf, G.A. 1981. Regulation of nitrogen metabolism and gene expression in fungi. *Microbiol. Rev.* 45:437-461.
- McGill, W.B., and C.V. Cole. 1981. Comparative aspects of organic C, N, S, and P cycling through organic matter during pedogenesis. *Geoderma* 26:267-286.
- North, M.J. 1982. Comparative biochemistry of the proteinases of eucaryotic microorganisms. *Microbiol. Rev.* 46:308-340.
- Payne, J.W. (ed.) 1980. *Microorganisms and nitrogen sources: Transport and utilization of amino acids, peptides, proteins, and relation substrates*. John Wiley and Sons, New York.
- Pinck, L.A., F.E. Allison, and M.S. Sherman. 1950. Maintenance of soil organic matter. II. Losses of carbon and nitrogen from young and mature plant materials during decomposition in soil. *Soil Sci.* 69:391-401.
- Shields, J.A., E.A. Paul, and W.E. Lowe. 1974. Factors influencing the stability of labeled microbial materials in soils. *Soil Biol. Biochem.* 6:31-37.
- Snyder, J.D., and J.A. Trofymow. 1984. A rapid accurate wet oxidation diffusion procedure for determining organic and inorganic carbon in plant and soil samples. *Commun. Soil Sci. Plant Anal.* 15:587-597.
- Turk, L.M., and C.E. Millar. 1936. The effect of different plant materials, lime and fertilizers on the accumulation of soil organic matter. *J. Am. Soc. Agron.* 28:210-324.
- Van Veen, J.A., W.B. McGill, H.W. Hunt, M.J. Frissel, C.V. Cole. 1981. Simulation models of the terrestrial nitrogen cycle. In F.E. Clark and T. Rosswall (ed.) *Terrestrial nitrogen cycles*. *Ecol. Bull. (Stockholm)* 33:25-48.